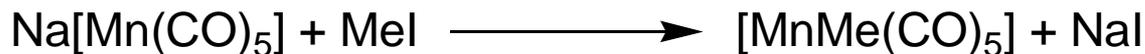
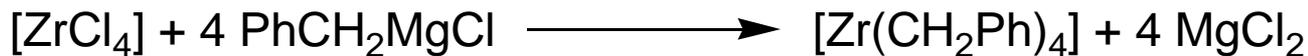


Alkyl Ligands

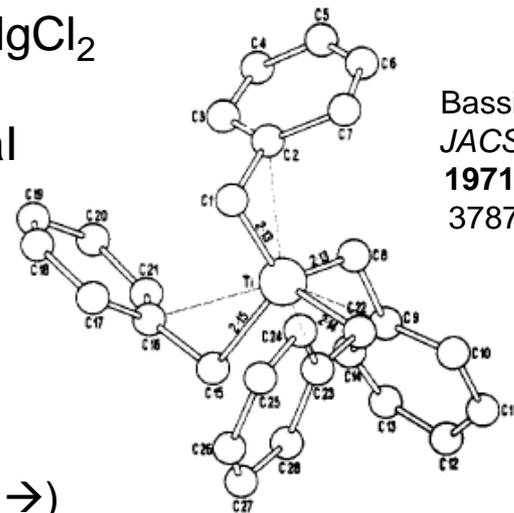


- **Not that many homoleptic TM alkyl complexes.**

$[\text{WMe}_6]$ = melts 30 °C

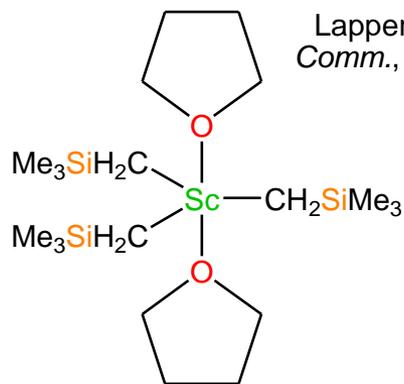
$[\text{TiMe}_4]$ = decomposes -40 °C

$[\text{TiBn}_4]$ = stable above room temperature (see crystal structure →)



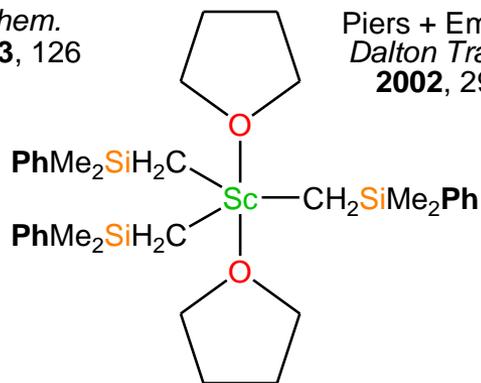
Bassi,
JACS,
1971,
3787

- **More sterically hindered complexes generally more stable.**



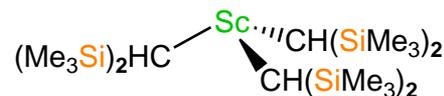
Lappert, *Chem. Comm.*, **1973**, 126

decomp > 0 °C



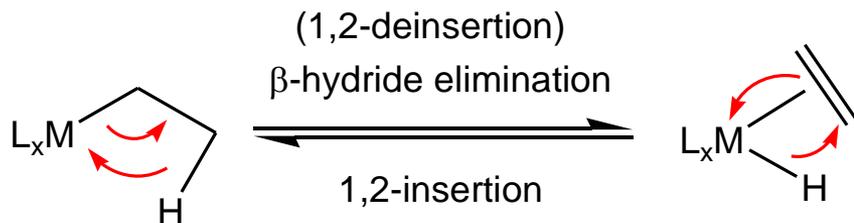
Piers + Emslie,
Dalton Trans.,
2002, 293

decomp > 40 °C



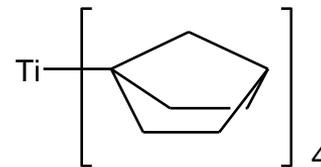
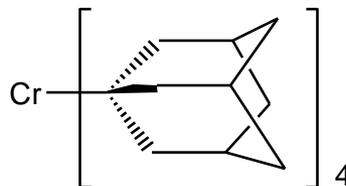
Alkyl Ligands

- Alkyl complexes far more stable if they lack β -hydrogens.



- Common alkyl groups unable to undergo β -hydride elimination:

- Me, CH_2Ph
- CH_2CMe_3 , CH_2SiMe_3 , $\text{CH}(\text{SiMe}_3)_2$
- CH_2CF_3
- 1-adamantyl, 6-norbornyl (see diagrams \rightarrow)



- Related ligands:

